

Chlorine Dioxide Complex Cleanser: A New Agent With Rapid Efficacy for Keratosis Pilaris

Matthew J. Zirwas MD^a and Jill Fichtel MD^b

^aOhio Contact Dermatitis Center, Dermatologists of Greater Columbus, Columbus, OH

^bDermatologists of Greater Columbus, Columbus, OH

ABSTRACT

Chlorine dioxide complex™ is a new molecule to dermatology that is a unique, non-toxic, broad spectrum anti-microbial and keratolytic compound. Chlorine dioxide has been used as an antiseptic in industrial settings for decades, primarily in water treatment facilities for municipal water supplies and food preparation. The compound has exceptional antiseptic properties with no known potential for development of resistance. It is a true keratolytic and anti-inflammatory, but is non-toxic to human tissue due to its unique mechanism of action. Chlorine dioxide's use in consumer products was previously limited because it is inherently an unstable molecule that had to be used quickly after it was produced. However, the recent development of a complexed form of chlorine dioxide that retains its antimicrobial and keratolytic activity has allowed the development of products (AsepticMD, Aseptic Plus, Nashville, TN) that take advantage of the properties of this unique molecule. Here we report a case series demonstrating its efficacy as a cleanser in keratosis pilaris.

J Drugs Dermatol. 2018;17(5):554-556.

INTRODUCTION

Chlorine dioxide (ClO₂) is a volatile, reactive gas at room temperature that was first synthesized in 1811. Early uses included as a pulp bleaching agent for paper making and as a disinfectant, but it was too expensive for routine industrial use until the 1940s. It has since been used in 5-10% of municipal water treatment facilities for water disinfection and deodorizing and is widely used as a food disinfectant.^{1, 2} Although ClO₂ gas is toxic at high concentrations by inhalation, when it is solubilized in water and applied to human tissue there is an absence of toxicity due to deactivation by intracellular defenses and the unique mechanism of action.³

When tested for antibacterial effectiveness against other antiseptics, such as bleach, hydrogen peroxide, iodophors, chlorhexidine, quaternary ammonium compounds, and others, ClO₂ is the most effective agent, both against typical organisms and against multidrug resistant Staph and Pseudomonas.⁴⁻⁶ It also has excellent activity against viruses, yeast and mycobacteria, as well as bacterial spores and biofilms.⁶⁻¹⁷ In addition, it is anti-inflammatory by neutralizing reactive oxygen molecules and cytokines and also acts as a true keratolytic, degrading both the inter- and intramolecular disulfide bonds that stabilize keratin.¹⁸⁻²⁰

While ClO₂ complex™ has numerous potential applications in dermatology, early experience has demonstrated exceptional efficacy in keratosis pilaris, primarily related to its keratolytic effects.

Report of Cases

Table 1 outlines the demographics, clinical findings, and response to therapy of patients treated with chlorine dioxide

complex wash (AsepticMD, Nashville, TN) in the authors' practice. All patients used the foaming facial cleanser and were instructed to wash once daily, gently rubbing the affected area for 5-10 seconds with a soft cotton cloth. No additional moisturizers, abrasion, or other interventions were used.

DISCUSSION

Keratosis Pilaris (KP) is a common complaint, affecting between 10% and 30% of the population. Onset is typically in childhood and although it frequently improves with age it does commonly persist into adulthood. Typical therapies include moisturizers containing keratolytics such as salicylic acid, lactic acid (or ammonium lactate) or urea. These treatments typically take several weeks to have an effect and must be used on an ongoing basis.^{21, 22} Additionally, more aggressive therapy with topical retinoids has been attempted, but can be limited by irritation and cost.²³ Finally, several lasers have been reported to be effective, but treatment is again limited by cost.²⁴⁻²⁹

Chlorine dioxide complex wash led to rapid, nearly complete resolution of keratosis pilaris in the reported patients. It presents several benefits over existing treatments, including low cost and ease of use – essentially all individuals use soap on a daily basis, while daily application of moisturizer is much less likely, especially in adolescents who have notoriously low compliance.

Unlike other keratolytics, ClO₂ is a highly specific oxidizer – it reacts with several specific amino acids in proteins, but does not react with or oxidize lipids, carbohydrates, or other organic

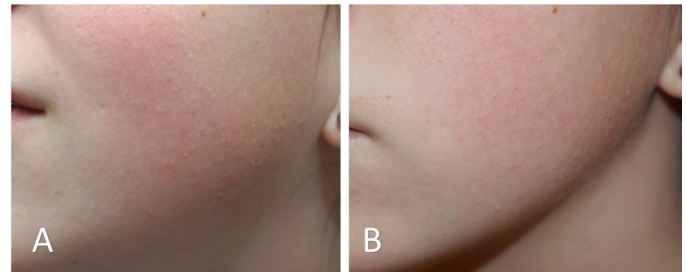
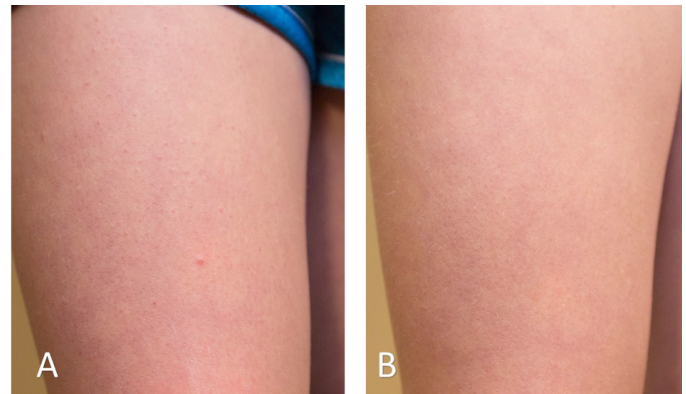
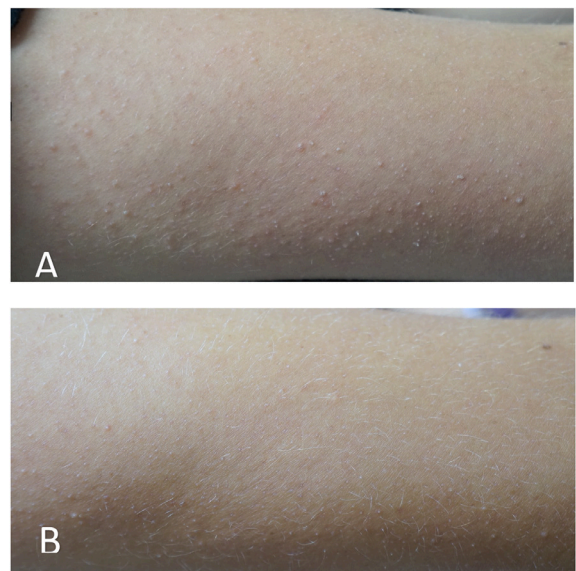
TABLE 1.**Clinical Findings and Result of Therapy with Chlorine Dioxide Complex Wash**

| Demographics | Involved area | Response to therapy |
|---------------|-----------------|--|
| 12 y/o female | Cheeks | Papules resolved in 3 days, minimal effect on erythema (Figure 1) |
| 11 y/o female | Anterior Thighs | Papules resolved in 2 days, minimal effect on erythema (Figure 2) |
| 13 y/o female | Posterior arms | Papules resolved in 2 weeks, minimal effect on erythema (Figure 3) |
| 28 y/o female | Posterior arms | Papules 90% resolved in 1 month, minimal effect on erythema |
| 20 y/o female | Cheeks | Papules resolved in 2 weeks, minimal effect on erythema |

molecules.³ Specifically, it reacts with cysteine, tyrosine, tryptophan, methionine, proline, hydroxyproline, and histidine, with most of its biologic activity coming from the reactions with cysteine, methionine, tyrosine, and tryptophan.³⁰⁻³⁴

This mechanism of action explains chlorine dioxide complex's rapid efficacy compared to traditional treatments for keratosis pilaris. Agents called "keratolytic," such as lactic acid, salicylic acid, benzoyl peroxide, ammonium lactate, and urea at concentrations less than 40%, do not have any direct effect on keratin – instead they hydrate the stratum corneum, allowing proteases to resume normal function and desquamation to proceed more normally than in dry conditions.³⁵ This has the obvious drawback of being slow, as it does not actually accelerate the rate of keratin removal compared to normal skin – it only returns it to being closer to a normal rate. ClO₂, on the other hand, directly attacks the cysteine residues in keratin as well as the inter- and intramolecular disulfide bonds between and within keratin chains, thus acting as a true keratolytic that directly softens keratin and reduces its cohesiveness.^{18, 19}

Chlorine dioxide's specificity for certain amino acids also explains its lack of toxicity. Non-specific oxidizers damage cells both internally by oxidizing proteins and externally by oxidizing lipids in cell membranes. ClO₂ does not damage cell membranes because it does not oxidize lipids. Then, when ClO₂ does enter cells, it rapidly reacts with the cysteine residues in glutathione, a key intracellular antioxidant peptide in both bacteria and mammalian cells. Because the reaction with cysteine is the most rapid chemical activity of ClO₂, as long as a cell has reserves of active glutathione with unreacted cysteine, the ClO₂ will not damage other cellular components.^{3,30} Obviously there is potential for ClO₂ toxicity to mammalian

FIGURE 1. Facial keratosis pilaris at baseline (A) and after 2 days (B) of using only chlorine dioxide complex wash once daily.**FIGURE 2.** Keratosis pilaris of the thigh at baseline (A) and after 2 days (B) of using chlorine dioxide complex wash once daily.**FIGURE 3.** Keratosis pilaris of the upper arm before (A) and after (B) two weeks of once daily use of chlorine dioxide complex wash.

cells if the concentration is high enough, and while this exact concentration has not been defined, it is much higher than the concentration of ClO₂ in use, giving ClO₂ a very wide

therapeutic window.³ Alternatively, bacteria have a substantially lower intracellular reserve of glutathione that is easily depleted, allowing chlorine dioxide to become cytotoxic.

Given its unique properties, there are a number of obvious uses for chlorine dioxide in dermatology, so one must wonder why it has not already been used in dermatology. The answer lies in the practical problem of low stability – ClO₂ degrades long before any product would reach the end user. One attempt to overcome this involved two bottle systems, in which two solutions that reacted to form ClO₂ were mixed together just before use – while these products have existed, especially as mouth rinses and wound care agents, the formulations possible were severely limited and they were quite inconvenient. This practical problem has been overcome by a unique process that complexes the chlorine dioxide into a stable compound that retains all of the useful properties of ClO₂. This chlorine dioxide complex, while giving significant flexibility in formulation, still must be formulated carefully to ensure it does not react with other components of the formulation.

DISCLOSURES

Dr. Zirwas has served as a consultant for Valeant, Sun Products, Medimetrix, Promius, Anacor, Exeltis, Genentech, Fitbit, and Smart Practice. He has been involved in the development of chlorine dioxide complex and is a part owner for AsepticMD.

Dr. Fichtel has an ownership stake in Strathspey Crown and Alphaeon, neither of which have any relationship with AsepticMD.

Chlorine dioxide complex was developed by Frontier Pharmaceutical and is patent pending.

REFERENCES

- US Department of Health and Human Services, Agency for Toxic Substances and Disease Registry: Toxicological Profile for Chlorine Dioxide and Chlorite. 2004.
- Fukayama, M.Y., et al., Reactions of aqueous chlorine and chlorine dioxide with model food compounds. *Environ Health Perspect*, 1986. 69:267-74.
- Noszticz, Z., et al., Chlorine dioxide is a size-selective antimicrobial agent. *PLoS One*, 2013. 8(11):e79157.
- Hinenoya, A., et al., Chlorine Dioxide is a Better Disinfectant than Sodium Hypochlorite against Multi-Drug Resistant *Staphylococcus aureus*, *Pseudomonas aeruginosa*, and *Acinetobacter baumannii*. *Jpn J Infect Dis*, 2015. 68(4):276-9.
- Tanner, R.S., Comparative Testing and Evaluation of Hard-Surface Disinfectants. *J Indust Micro*, 1989. 4:145-154.
- Herczegh, A., et al., Comparing the efficacy of hyper-pure chlorine-dioxide with other oral antiseptics on oral pathogen microorganisms and biofilm in vitro. *Acta Microbiol Immunol Hung*, 2013. 60(3):359-73.
- Patel, M., J. Ebonwu, and E. Cutler, Comparison of chlorine dioxide and dichloroisocyanurate disinfectants for use in the dental setting. *SADJ*, 2012. 67(7):364, 366-9.
- Uludamar, A., A.G. Ozyesil, and Y.K. Ozkan, Clinical and microbiological efficacy of three different treatment methods in the management of denture stomatitis. *Gerodontology*, 2011. 28(2):104-10.
- Wei, M.K., et al., Plasma membrane damage to *Candida albicans* caused by chlorine dioxide (ClO₂). *Lett Appl Microbiol*, 2008. 47(2):67-73.
- Mohammad, A.R., et al., Clinical and microbiological efficacy of chlorine dioxide in the management of chronic atrophic candidiasis: an open study. *Int Dent J*, 2004. 54(3):154-8.
- Morino, H., et al., Effect of low-concentration chlorine dioxide gas against bacteria and viruses on a glass surface in wet environments. *Lett Appl Microbiol*, 2011. 53(6):628-34.
- Vicuna-Reyes, J.P., J. Luh, and B.J. Marinas, Inactivation of *Mycobacterium avium* with chlorine dioxide. *Water Res*, 2008. 42(6-7):1531-8.
- Taylor, G.R. and M. Butler, A comparison of the virucidal properties of chlorine, chlorine dioxide, bromine chloride and iodine. *J Hyg (Lond)*, 1982. 89(2):321-8.
- Chen, Y.S. and J.M. Vaughn, Inactivation of human and simian rotaviruses by chlorine dioxide. *Appl Environ Microbiol*, 1990. 56(5):1363-6.
- Gagnon, G.A., et al., Disinfectant efficacy of chlorite and chlorine dioxide in drinking water biofilms. *Water Res*, 2005. 39(9):1809-17.
- Chauvet, C.P., et al., Chlorine dioxide inactivation of *Cryptosporidium parvum* oocysts and bacterial spore indicators. *Appl Environ Microbiol*, 2001. 67(7):2993-3001.
- Hernandez, A., M. Carrasco, and V. Ausina, Mycobactericidal activity of chlorine dioxide wipes in a modified prEN 14563 test. *J Hosp Infect*, 2008. 69(4):384-8.
- D. B. Bass, J.B.S., The Action of Chlorine Dioxide on Wool. *J Soc Dyers Colorists*, 1950. 66(11):583-587.
- Bragulla, H.H. and D.G. Homberger, Structure and functions of keratin proteins in simple, stratified, keratinized and cornified epithelia. *J Anat*, 2009. 214(4):516-59.
- Kenyon, A.J., S.G. Hamilton, and D.M. Douglas, Controlled wound repair in guinea pigs, using antimicrobials that alter fibroplasia. *Am J Vet Res*, 1986. 47(1):96-101.
- James WD, B.T., Elston D., *Andrew's Diseases of the Skin: Clinical Dermatology*. 11th ed. 2011, London: Saunders.
- Novick, N.L., Practical management of widespread, atypical keratosis pilaris. *J Am Acad Dermatol*, 1984. 11(2 Pt 1):305-6.
- Gerbig, A.W., Treating keratosis pilaris. *J Am Acad Dermatol*, 2002. 47(3):457.
- Schoch, J.J., et al., Successful Treatment of Keratosis Pilaris Rubra with Pulsed Dye Laser. *Pediatr Dermatol*, 2016. 33(4):443-6.
- Vachiramon, V., et al., Fractional Carbon Dioxide Laser for Keratosis Pilaris: A Single-Blind, Randomized, Comparative Study. *Biomed Res Int*, 2016. 2016:1928540.
- Ibrahim, O., et al., Treatment of keratosis pilaris with 810-nm diode laser: a randomized clinical trial. *JAMA Dermatol*, 2015. 151(2):187-91.
- Saelim, P., et al., Long-pulsed 1064-nm Nd:YAG laser significantly improves keratosis pilaris: a randomized, evaluator-blind study. *J Dermatolog Treat*, 2013. 24(4):318-22.
- Lee, S.J., et al., Combination of 595-nm pulsed dye laser, long-pulsed 755-nm alexandrite laser and microdermabrasion treatment for keratosis pilaris. *J Dermatol*, 2012. 39(5):479-80.
- Park, J., et al., A Pilot Study of Q-switched 1064-nm Nd:YAG Laser Treatment in the Keratosis Pilaris. *Ann Dermatol*, 2011. 23(3):293-8.
- Ison, A., I.N. Odeh, and D.W. Margerum, Kinetics and mechanisms of chlorine dioxide and chlorite oxidations of cysteine and glutathione. *Inorg Chem*, 2006. 45(21):8768-75.
- Tan, H.K., W.B. Wheeler, and C.I. Wei, Reaction of chlorine dioxide with amino acids and peptides: kinetics and mutagenicity studies. *Mutat Res*, 1987. 188(4):259-66.
- Loginova IV, R.S., Kuchin AV, Oxidation by Chlorine Dioxide of Methionine and Cysteine derivatives to sulfoxide. *Chem Nat Compd*, 2008. 44:752-754.
- Napolitano, M.J., et al., Chlorine dioxide oxidations of tyrosine, N-acetyltyrosine, and dopa. *Chem Res Toxicol*, 2005. 18(3):501-8.
- Stewart, D.J., et al., Kinetics and mechanisms of chlorine dioxide oxidation of tryptophan. *Inorg Chem*, 2008. 47(5):1639-47.
- Vyumvuhore, R., et al., Vibrational spectroscopy coupled to classical least square analysis, a new approach for determination of skin moisturizing agents' mechanisms. *Skin Res Technol*, 2014. 20(3):282-92.

AUTHOR CORRESPONDENCE

Matthew J. Zirwas MD

E-mail:..... matt.zirwas@gmail.com